ECTFE Halar ®

Thermoplastic High-tech Coating

- Excellent resistance to chemicals
- First-rate electrical properties
- Superior mechanical properties
- ☆ Temperature proof -70℃ bis +150℃
- Low rate of permeation
- Variant for use with foodstuffs
- Good weatherproofness
- ✤ Water absorption < 0,1%</p>
- Inflammable to UL94 V-0
- Resistant to x-rays up to 200 MRad
- ✤ Easily machineable
- ✤ High level coating thickness 400 1000µ

The guidelines of DIN 29051 apply to the finishing of the articles to be coated.

Resistance to chemicals

HALAR provides an superior resistance to chemicals and an excellent diffusion barrier. It remains practically unaffected when in contact with the majority of corrosive chemicals used in industry, e.g. concentrated mineral and oxydizing acids, alkalis, metal etching products, liquid oxygen, as well as virtually all organic solvents, except the hot amines (e.g. aniline, dimethylamine).

HALAR swells when exposed to natrium and potassium compounds. The degree of swelling depends on the length of exposure and temperature.

HALAR and other fluorpolymers can also swell when in contact with certain halogenised solvents. This process does not normally its usage. When the solvent has evaporated and the surface is dry, the mechanical properties return to their original values, proving that no lasting chemical degradation has taken place.

Further information about the chemical stability of HALAR can be found in the following table 1.1 (green appendix) and 1.2 (yellow appendix).

Electrical properties

HALAR possesses a high volume and surface resisitivity and a very low loss factor. HALAR's dielectric constant is 2.6 and remains stable over a wide temperature and frequency range.

HALAR has a high dielectric strength (ASTM D149): In HALAR coatings of 1.0 mm thickness this is approx. 35 kV/mm (see Table 2).

HALAR can also be modifed to provide antistatic properties with a resistance of between 10⁹ Ω und 10⁶ Ω .

Table 2: Electrical properties					
Electrical properties	Testing method ASTM	Units	Value		
dieletric constant	D-150 50 Hz 10 ³ Hz 10 ⁶ Hz	1	2,6 2,5 2,5		
dielectric loss factor	D-150 50 Hz 10 ³ Hz 10 ⁶ Hz	1	> 0,0009 0,0017 0,017		
transverse or volume resisitivity	D-257	Ω cm	10 ¹⁵		
surface resistivity	D-257	Ω	> 10 ¹⁵		
dielectric strength	D-149 - 0,025 mm film - 1,0 mm coating	KV/mm	80 theor. value 35 prac. value		
Resistance to arcing	D-495	sec.	135		

Mechanical properties

HALAR is a hard, shockproof polymer, whose properties remain practically constant over a wide temperature range. Its performance at low temperatures down to -80°C, especially its resistance to impact is outstanding. Additionally HALAR displays good tensile strength and resistance to abrasion. (see table 3).

Table 3: Mechanical properties				
Properties	Testing method ASTM/DIN	Units	Value	
tensile strength	D-638 at point of failure yield strain	MPa MPa	50 32	
elongation at failure	D-638	%	200	
modulus of elasticity	D-790	MPa	1700	
modulus in tension	D-638	MPa	1700	
flexural strength	D-790	MPa	43	
impact strength when notched IZOD	D-256	+23℃ - 40℃	Does not tear 2 – 3	
hardness	D-785 53505	Rockwell Shore D	R-93 75	
indentation hardness	53456	132/60 MPa	55-65	
abrasion resistance	D-1242 D-1044	Armstrong cc Taber 500 U gr 1000 U gr	0,3 loss of volume 0,002 0,005	
friction coefficient	ASTM D-1894		0, 7 – 0,8	

Thermal properties / Inflammability / UL

HALAR is s heated uitable for constant usage at a temperature of +150°C. A short period of exposure of up to +180°C is possible.

HALAR takes on a brownish colour when exposed for a longer time to the temperatures given above.

HALAR's nonflammability is one of its excellent properties. When exposed to fire HALAR carbonises while most of the other fluoroplastics melt and drip. The carbonisation process stops immediately the flames are removed.

HALAR has been assessed as UL94-VO material by Underwriters Laboratories (UL) als UL94-VO Material.

Production of fumes in case of fire is lower with HALAR than with most other thermoplastics.

Table 4: General thermal properties of HALAR					
Properties	ASTM method	Unit	Value		
coefficient of linear expansion	D-696 -30℃ bis +30℃	1	5 X 10 ⁻⁵		
thermal conductivity	40℃ 40℃	Joule/m h℃ Kcal/m h℃	556 0,13		
Resistance to thermal deformation under pressure 4,6 kp/cm ² 18,6 kp/ cm ²		С С	115 76		
Melting point	DSC	c	245		
embrittlement at low temperatures	D-746	C	-80		
max. constant usage temperature without strain.		С С	150 180 momentary		
nonflammability	D-635	UL 94 vertical 0,18 mm	V-0 does not drip does not burn		
inflammability L0I (limit oxygen index)	D 2863	%0 ₂	60		

Permeation / Diffusion barrier

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HALAR displays a very low permeability to steam and oxygen, as well as various other gases. Absorption of humidity is less than 0,1 %. HALAR can only be sterilised to a limited extent.





Contact with foodstuffs

HALAR type 8014 complies with the requirements of Food-Additive Regulations, use B, set down in 21 C.F.R. 176.170 ©, Tab. 2.

Weatherproofness

HALAR displays very little change in its properties and appearance when used outdoors. Concentrated weather tests have proved a considerable stability of the polymer. Essential properties are scarcely affected even after 1000 hours in weather-ometer with xenon light.

Resistance to radiation

HALAR is one of the polymers most resistant to radiation, e.g.HALAR retains its good properties even after cobalt 60 radiation of 200 Mrad.

Long-term reactions

HALAR's life-expectancy by constant use at 150 $\ensuremath{\mathbb{C}}$ is approximately 20 years (limitation:

50% tensile strength, however, still fulfilling its prime function).

HALAR produces many times smoother microsurfaces than, for example PVDF, PVC, PP and is water repellent.

The degree of crystallinity and thus the mechanical and thermal properties remain permanently stable.

Application

Substrate-preparation:

- degreasing
- corundium blasting, scale of cleanness Sa3

Priming:

Coating is carried out by electrostatic powder spraying on to the primer coat heated to 360° - 390° . There are two different priming syst ems available:

- thermoplastic primer coat, black or

- thermoset primer coat, green/brown

Coating

- fusing of several coats at a substrate temperature of +360°C to +390°C.
- a thickness of 100 200my can be applied at each coating. This means that 4 6 coats must be separately fused, in order to achieve the standard layer thickness for pretection against chemicals.

Retouching of suspension points

- re-melting, where necessary with the addition of HALAR

Quality control

- measurement of layer thickness with Fischer Dualscope
- Pinhole testing with Fischer poroscope 2,5 25 kV

Essential specifications of HALAR 6014

Property	Unit	6014
Mechanical properties Hardness tensile strength tensile strength at break elongation at tear flexural module impact strength Izod, with notch, +23℃ Izod, with notch, -40℃	Shore D MPa MPa % MPa J/m J/m	75 32 45 325 1520 kein Br uch 48
Elelctrical properties		
dielectric strength coating 800 μ dielectric constant at 10 ³ Hz at 10 ⁶ Hz loss factor at 10 ³ Hz at 10 ⁶ Hz	kV/mm 1 1	34,6 2,45 2,57 0,0017 0,017
Resistance to chemicals		
Sulphuric acid 98 % Concentrated hydrochloric acid Hydrofluoric acid Sodium hydroxide, 50 %		no effects of corrosion no effects of corrosion no effects of corrosion no effects of corrosion
Inflammability		
Oxygn index, 1,6 mm UL 94 vertical, 0,18 mm Nonflammability and dischagre of smoke	(fumes?)	min. 52 94 V-O i.O.
Thermal properties		
Melting point max. temperature for usage	С С	245 -70/+150
Other specifications		
Resistance to radiation weatherproofness Specific gravity Absorption of humidity	MRad gr/cm ³ %	200 good 1,68±0,05 <0,1
Application		
Fusing temperature	C	+260 bis +300
Eposint AG / K. Gantenbein/ua		

Table 1.1 Resistance to chemicals

	Chemical	Temp °C	Weight increase, %	Effect on elasticity of elongation	Effect on elongation at tear
Effect on properties	Mineral acids (*4)			Ŭ	
	sulphuric acid 78%	23 121	<0.1 <0.1	U U	U U
*1 There was no evidence of tears due to strain(tension) on	Hydrochloric acid 37%	23	<0.1	U	U
the 2.29mm test objects in any		75-105	0.1	U	U
of the experiments according to ASTM D1693.	Hydrochlori acid 60%	23	<0.1	U	U
	Chlorosulfone acid	23	0.1	U	U
* 2 Key :U = unimportant A = reduction by 25- 50%	Oxydising acids	22	-0.1		
B = reduction by 50-	Nitric acid 70%	23 121	<0.1 0.8	Δ	0 C
C = reduction by >75%	Chromic acid	23	<0.0	Ŭ	U
*3 This data is based on 11		111	0.4	U	U
day tests on 2.29mm thick test	Nitrohydrochloric	23	0.1	U	U
tears). The loss in weight was measured within 15 minutes	aciu	75-105	0.5	U	U
after removal from the testing fluid. The test objects were then wrapped in foil and sealed	Anorganic salts Ferrous chloride	00	0.4		
in glass jars to minimise the	25% Zinc chloride 25%	23	0.1	U	U
release of solvent. The tests for		1 03	0.1	U	U
hours after their removal (test objects according to ASTM D1693) *4 HALAR contains application additives, which darken in colour when exposed for longer periods to certain highly concentrated acids (e.g.		23	<0.1	U	U
	Copper chloride 25%	104	<0.1	U	U
		23	<0.1	U	U
	Anonyonia allastia	1 03	<0.1	U	U
	Sodium hydroxide	23	<0.1	U	U
Suprune and hittle acid.)		121	<0.1	U	U
	Ammomnium hrdroxide 28%	23	<0.1	U	U
		66	0.2	U	U
	Halogens				
	Bromine	23	1.4	U	U
	Solvents				
	Aliphatic compounds	23	0.1	U	U
	Hexane (solvents)	54	1.4	A	U
	Chemical	Temp	Weight	Effect on	Effect on

		°C	increase, <mark>%</mark>	elasticity of elongation	elongation at tear
Effect on properties	Isooctane	23	<0.1	U	U
		116	3.3	А	U
*1 There was no evidence of	Aromatic compounds				
the 2 29mm test objects in any	Benzene	23	0.6	U	U
of the experiments according to		74	7.0	С	U
ASTM D1693.	Methylbenzene	23	0.6	U	U
		110	8.5	С	U
* 2 Key :U = unimportant A = reduction by 25-	Ketones				
50% B = reduction by 50-	Acetone	23	0.1	U	U
75% C = reduction by >75%		52	4.0	В	U
*3 This data is based on 11	methylethylketone	23	1.0	U	U
day tests on 2.29mm thick test		79	6.0	С	U
tears) The loss in weight was	Methylisobutylketone	23	0.5	А	U
measured within 15 minutes after removal from the testing		115	9.0	С	U
fluid. The test objects were	Acids				
then wrapped in foil and sealed	Acetic acid pure	23	0.2	U	U
release of solvent. The tests for tension followed within 24		110	3.5	С	U
hours after their removal (test	Ester				
objects according to ASTM D1693)	Ethylacetate	23	0.2	U	U
		71	6.5	В	U
	Butylacetate	23	0.2	U	U
*4 HALAR contains application		121	10.5	C	U
colour when exposed for longer periods to certain highly concentrated acids (e.g.	Dimethylphatat	23	<0.1	U	U
		121	3.5	A	U
sulphuric and nitric acid.)	Amines				
	Diethylamine	23	0.2	U	U
	2.00.19.00.000	118	?	2	2
	Aniline	23	<0.1	Ŭ	Ū
		121	?	?	?
	Standard plastic solvents				-
	Dimethylformamide	73	2.0	А	U
	-	250	7.5	С	U
	Dimethylsulfoxide	73	0.1	U	U
		250	3.0	U	U

	Chemical	Temp °C	Weight increase, %	Effect on elasticity of elongation	Effect on elongation at tear
	(solvents)			-	
Effect on properties	Functional aromatic compounds				
	Aniline	23	<0.1	U	U
		121	1.4	А	U
*1 Thore was no ovidence of	Benzaldehyde	23	0.2	U	U
*1 There was no evidence of tears due to strain(tension) on		121	10.4	С	U
the 2.29mm test objects in any	Chlorbenzol	23	0.9	С	U
of the experiments according		121	19.5	U	U
to ASTM D1693.	Dimethylphtalate	23	<0.1	U	U
		121	3.5	А	U
*2 Key:U = unimportant	Nitrobenzol	23	0.2	U	U
A = reduction by 25-		121	11.5	С	U
50% B – reduction by 50-	Chlofied solvents				
75%	Chloroform	23	4.5	А	U
C = reduction by >75%	Methylenchloride	23	8.0	В	U
*3 This data is based on 11		40	9.0	С	U
day tests on 2.29mm thick test	Trichlorethylene	23	5.0	В	U
tears). The loss in weight was		85	16.5	С	U
measured within 15 minutes	Perchlorethylene	23	1.0	U	U
after removal from the testing		121	29.0	С	U
fluid. The test objects were	Dichlorethylene	23	1.0	С	U
sealed in class jars to		85	9.5	U	U
minimise the release of	R113	23	0.4	U	U
solvent. The tests for tension		49	2.0	U	U
followed within 24 hours after	chlorbenzol	23	0.9	А	U
according to ASTM D1693)		121	19.5	С	U
······································	Alcohols				
*4 HALAR contains	methanol	23	0.1	U	U
darken in colour when		60	0.4	А	U
exposed for longer periods to	buthanol	23	0.1	U	U
certain highly concentrated		118	2.0	А	U
acids (e.g. sulphuric and nitric	Ethers				
acid.)	ethylether	23	0.9	U	U
	dioxane	23	0.9	U	U
		102	16.0	С	U
	Propylenoxide	23	6.0	С	U
	tetrahydrofurane	23	4.5	В	U
		63	11.0	С	U